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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.004 Å R factor = 0.064 wR factor = 0.170 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N-{3-Cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-1*H*-pyrazol-5-yl}acetamide

The title compound, $C_{13}H_7Cl_2F_3N_4O$, has normal bond lengths and angles. The crystal packing is stabilized by intermolecular $N-H\cdots N$ hydrogen bonds.

Comment

Reaction of 2,6-dichloro-4-trifluoromethylamine with a suspension of nitrosyl sulfuric acid, followed by reaction with a solution of ethyl 2,3-dicyanopropionate in acetic acid, gave 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-pyrazole, which then reacted with acetic anhydride to give the title compound, (I) (Fig. 1).



The title compound is used in the synthesis of insecticides, such as 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)-phenyl]-4-(trifluoromethylthio)pyrazole, 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-4-(trifluoromethyl)pyrazole and 5-amino-3-cyano-1-[2,6-dichloro-4-(trifluoromethyl)pyrazole (Hatton *et al.*, 1993).



© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved View of (I), with the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level.

organic papers

All bond lengths and angles in (I) are normal (Table 1). The crystal packing (Fig. 2) is stabilized by the weak intermolecular $N-H \cdots N$ hydrogen bonds (Table 2).

Experimental

Compound (I) was synthesized and purified according to the method of Hatton et al. (1993). Single crystals suitable for X-ray data collection were obtained by slow evaporation of an ethyl acetatetoluene (1:1) solution (m.p. 477-478 K). Spectroscopic analysis, IR (KBr, ν cm⁻¹): 3446, 2383, 1698, 1649, 1520; ¹H NMR (CDCl₃, δ , p.p.m.): 7.82 (s, 2H), 7.08 (s, 1H), 7.00 (s, 1H), 2.13 (s, 3H).

 $D_x = 1.559 \text{ Mg m}^{-3}$

Cell parameters from 2156

Mo $K\alpha$ radiation

reflections

 $\theta = 2.6 - 25.2^{\circ}$ $\mu = 0.46~\mathrm{mm}^{-1}$

T = 298 (2) K

 $R_{\rm int} = 0.025$ $\theta_{\rm max} = 25.3^\circ$

 $h = -22 \rightarrow 20$

 $k = -10 \rightarrow 8$

 $l = -18 \rightarrow 22$

Block, colorless

 $0.44\,\times\,0.16\,\times\,0.13~\mathrm{mm}$

2768 independent reflections

 $w = 1/[\sigma^2(F_o^2) + (0.079P)^2]$

+ 4.3533P] where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} = 0.003$

 $\Delta \rho_{\rm max} = 0.65 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.52 \text{ e} \text{ \AA}^{-3}$

2373 reflections with $I > 2\sigma(I)$

Crystal data

C13H7Cl2F3N4O $M_r = 363.13$ Monoclinic, C2/c a = 18.8282 (6) Å b = 8.6541 (3) Å c = 19.0766 (6) Å $\beta = 95.583(1)^{\circ}$ $V = 3093.62 (18) \text{ Å}^3$ Z = 8

Data collection

Bruker SMART APEX areadetector diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Bruker, 2002) $T_{\min} = 0.824, T_{\max} = 0.943$ 7655 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.170$ S = 1.142768 reflections 209 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Selected geometric parameters (Å, °).

1.130 (5) 1.330 (5)
1.330 (5)
1.497 (5)
1.364 (4)
1.383 (5)
1.441 (5)
119.2 (2)
121.4 (3)
120.5 (3)
119.9 (3)





Table 2 Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$N3 - H3 \cdots N4^i$	0.86	2.22	3.060 (4)	166
Summatry and (i)	. 1			

Symmetry code: (i) x, 1 + y, z.

All H atoms were initially located in a difference Fourier map and were placed in geometrically idealized position and constrained to ride on their parent atom, with C-H distances in the range 0.95-1.00 Å and $U_{iso}(H) = 1.2_{eq}(C)$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL (Bruker, 2002); software used to prepare material for publication: SHELXL97.

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References

Bruker (2002). SMART, SAINT, SADABS and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

Hatton, L. R., Bunain, B. G., Hawkins, D. W., Parnell, E. W., Pearson C. J. & Roberts, D. A. (1993). US Patent No. 5 232 940.

Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.